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Secker Street 29
Hamburg. Dec. 1. 1894

Dear Sir

Many thanks for your letter and for satisfying me that Maxwell had ~~expressly~~ given equations for calculating the rate at ^{which} the equalization of pressure in different directions about the same point in a gas, takes place. This second paper I had missed, while reading the first.

In regard to your saying, you should like to know what I have to say to Maxwell's objection to a suggestion he attributes to Sir W. Thomson - I may remark on the fact of having considered this point in my second Phil Mag paper of Nov. 1877, page 370 etc.

The wording of my mode of surmounting the difficulty may be clearer somewhat in my doctorate "Dissertation" and possibly the following translation of the passage may facilitate: - viz

"In an article in the Ency. Brit 1875 (a Scientific Notes Vol 2, p. 476) Maxwell raises the objection that gross matter must be made white hot by the encounters of the other atoms. Maxwell bases ~~the~~

this objection on the theorem that equilibrium of temperature between any atoms or molecules demands ^{we} the mean energy of translatory motion of all should be equal same. Now the pressure is proportional to the product of the mean energy of translatory motion (call this L), into the number of atoms N , in the unit of volume. If then the mean translational energy of an other atom were equal to that of a molecule of the atmosphere (say) ^{latter}; which we can determine by calculation; then N (number of atoms in unit of volume) must have an enormous value for the other, in order to explain the pressure of gravity. Now Maxwell says: we are tolerably certain that N for the other is small compared with the value of N for gross matter - say for the atmosphere. From this Maxwell concludes that in order to explain the pressure of gravitation, ~~the~~ L must be assumed enormously great for the other. And from the theorem that for thermal equilibrium, L must be the same for all molecules, it would follow that L would assume a value for the molecule of gross matter far greater than that which we find for gases. In other words, that the whole of gross matter would be raised to a white heat by the collisions of other atoms.

But, even independently of the above theorem for thermal equilibrium being still disputed in relation to molecules of very different dimensions - it seems to me that

most certainly no cogent reason exists for ^{Maxwell's} ~~the~~ assumption,
of N being smaller ^{in value} for the æther than for gaseous
matter. One can in fact conceive the æther atoms
to have any finite size however small; then ~~can~~ ^{can} one
imagine a number of them in unit of volume, ^{at least,}
consistent with an enormously long mean path. (p. 10).

Such is the translation of the means offered for
surmounting Maxwell's difficulty, or objection.

Yours very truly
S. Toliver Preston

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Prof. G. F. Fitzgerald F.R.S.