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23 Ph Parade
Cambridge
20th Jan 95

Dear Mr Fitzgerald

I must confess that I don't see how you get the radius of the drop 10^{-5} . The density of electrification if one in every 10^7 were charged would be $\frac{10^{14}}{3 \cdot 10^{10}} = \frac{10^4}{3}$ which is 10^7 times the actual charge per c.c. that I get.

The theory I have been working upon is that if E is the charge a stable position is obtained when

$$\frac{E^2}{8\pi a^4} = \frac{2T}{a} \quad a = \text{radius} \ \& \ T \text{ the surface}$$

tension. But I find the charge on each is about 10^{-11} so this would give a radius of $\frac{1}{3} 10^{-8}$ which is far too small.

The thing that I have not done well yet is to get the size of the drops but I am perfectly certain that for a charge $2.3 \cdot 10^{-3}$ per c.c. the wt of the cloud at 11° is $18.4 \cdot 10^{-6}$. If I can only get a good determination, experimentally of the size of

the drop the thing could be tested then by the formula, but at present I can only get the drop between $2 \cdot 10^{-5}$ & $5 \cdot 10^{-5}$.

I find I can't get it on before the Philosophical Society here before Feb 8th, as I want to show some expts & the meeting would have to be at the Cavendish, so that I will have another week to do experiments & may probably get some thing out of the hydrogen cloud.

Is there any other effect due to the charged molecules outside the drop that could affect the equilibrium? I don't think the formula $\frac{E^2}{8\pi a^4} = \frac{2\pi}{a}$ would explain the size of the drops.

Yours sincerely
John S. Townsend

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topography - We also propose to
 have another lecture on Architecture
 making a course of 5 in all -

(Faint handwritten notes and calculations, possibly including a signature)

H. F. Fitzgibbon } F.R.S. etc.

4/3 *(with various scribbles and numbers)*

(Vertical list of numbers and symbols)
 4.6
 4.1 = .01 x 6 =
 = 4 x 10 x 2.3 x 10⁻⁶
 3/4 = 2/2 = 1/1.5

$$\frac{2T}{r} = \frac{1}{8\pi} \frac{Q^2}{r^3}$$

$$Q^2 = 16\pi r^3 \cdot 80$$

$$= 50 \times 80 \times \sqrt{3}$$

$$= 4000 \times \sqrt{3}$$

$$Q = \cancel{62} \times 10 \quad 62 \times r^{3/2}$$

$$r = 10^{-5}$$

$$r^{3/2} = 10^{-15/2}$$

$$= \sqrt{10} \times 10^{-16/2}$$

$$= 3 \times 10^{-8}$$

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$$Q = 186 \times 10^{-8}$$

$$= 1.86 \times 10^{-4}$$

$\therefore 10^7$ ions per drop.

So would hardly hold two mol. together.

There is a much smaller amount than to a much smaller amount of impurity drops. would be required to keep up.

$$Q = 10^{-11} = 62 \times r^{3/2}$$

$$r^{3/2} = \frac{1}{62} \times 10^{-11}$$

$$r^3 = \frac{10^{-22}}{4000} = \frac{10^{-24}}{40 \times 10^8}$$

$$r = \frac{10}{3.3} = 3 \times 10^{-9}$$

$$62^2 = 4000$$

3×10^{-9}